

Reaction of α -carbanion of imines with *N*-tosylimines: a facile route to β -aminoaldehydes and 1,3-diamines

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Deprotonation of aldimines by LDA, followed by trapping of the resulting carbanion with *N*-tosylimines and hydrolysis or reduction provides a convenient access to β -aminoaldehydes or 1,3-diamines.

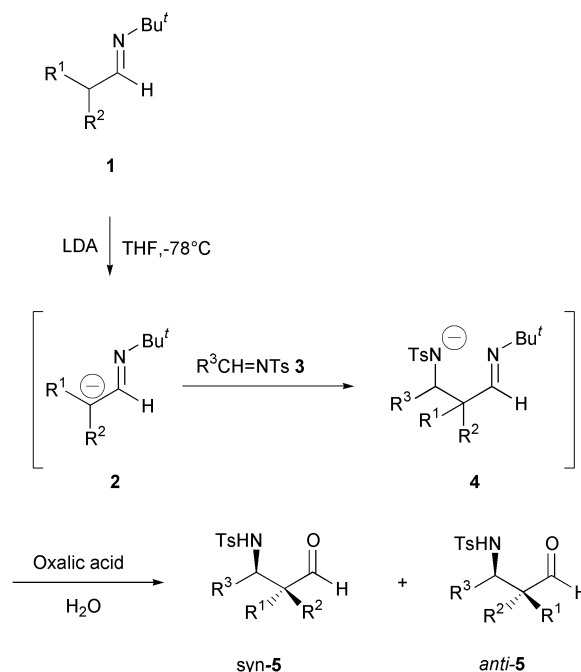
Introduction

Recently, the use of imines as starting materials in the synthesis of nitrogen-containing compounds has attracted a lot of interest from synthetic chemists.¹ A number of reactions of imines, such as aziridination,² alkylation, aldol reaction, hetero-Diels–Alder reaction, have been well documented.² All of these reactions utilize the carbon of the C=N double bond as an electrophile or the C=N double bond as dienophile. There are only a few examples of the use of α -carbanions derived from imines,³ although the reactions of α -carbanions of carbonyl compounds, an imine analog, are important in organic synthesis.⁴ As a program aimed at the applications of imines,^{2a,g,5} we have studied the formation of carbanions from imines and their subsequent reactions. Now we report the deprotonation of imines and the reaction of the thus formed α -carbanion with *N*-tosylimines to provide β -aminoaldehydes and 1,3-diamines.⁶

Results and discussion

In the presence of LDA, deprotonation of imines **1** gave rise to α -carbanions of imines **2**, which reacted with *N*-tosylimines **3** to afford β -amino imines **4**. Direct hydrolysis of amino imines **4** with oxalic acid delivered β -aminoaldehydes **5** (Scheme 1). The results are showed in Table 1.

This is a one-pot reaction. All aldimines **1** reacted with *N*-tosylimines **3** to give rise to the intermediates, β -aminoaldehydes **4**, which allowed direct hydrolysis. The β -aminoaldehydes were provided in good yield after three-step reactions. The substituent R¹ can be H and Me and R² can be



Scheme 1

Table 1 Synthesis of β -aminoaldehydes from aldimines **1** and **3**^a

Entry	R ¹	R ²	R ³	Product (yield %) ^b	syn : anti ^c
1	H	Me	4-ClC ₆ H ₄	5a (61)	40 : 60
2	Me	Me	4-ClC ₆ H ₄	5b (72)	
3	H	Et	Ph	5c (81)	23 : 77
4	H	Et	4-MeOC ₆ H ₄	5d (79)	26 : 74
5	H	Et	4-ClC ₆ H ₄	5e (80)	25 : 75
6	H	Pr	Ph	5f (63)	25 : 75
7	H	ⁱ Pr	Ph	5g (82)	8 : 92
8	H	ⁱ Pr	4-MeOC ₆ H ₄	5h (73)	22 : 78
9	H	ⁱ Pr	4-ClC ₆ H ₄	5i (83)	8 : 92

^a All reactions were carried out with the ratio of LDA : imine : *N*-tosylimine = 1.6 : 1.5 : 0.75. ^b Isolated yields based on the *N*-tosylimine. ^c Determined by 300 MHz ¹H-NMR.

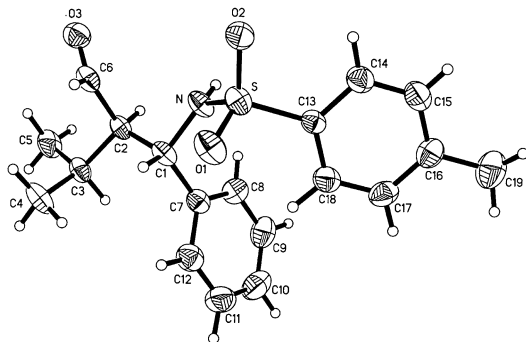
linear and branched alkyl group. The stereoselectivity of the reaction is usually low, but in some cases it is high (Entries 7 and 9). A change of solvent from THF to toluene had little influence on the stereochemistry outcome. The presence of additives, such as molecular sieves, LiClO₄, also did not change the *syn*–*anti* ratio of products. The stereochemistry of products was determined from the *J* value of the aldehyde proton from ¹H-NMR and confirmed further by X-ray diffraction analysis of **5g** (Fig. 1).⁷

In order to trap the carbanion intermediate **2**, the use of activated imine is crucial. The imines with phenyl and diphenylphosphinoyl groups as substituent on nitrogen failed to react with carbanions **2** to give any product. *N*-Tosylimines, with either electron-donating or electron-withdrawing substituents on the phenyl ring give the desired products. Enamines have been used widely as a synthon of α -carbanions of carbonyl compounds,⁸ however, the reaction of enamines derived from cyclohexanone and piperidine with PhCH=NTs or PhCH=NPh-TMSCl did not provide the desired products. The choice of base is also important.^{3e} In the presence of LDA, aldimines

Table 2 Synthesis of 1,3-diamines from aldimines **1** and **3**^a

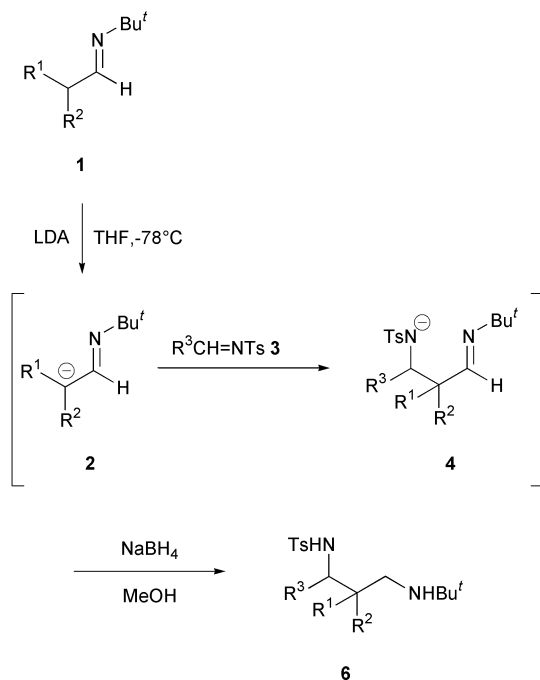
Entry	R ¹	R ²	R ³	Product (yield %) ^b
1	H	C ₂ H ₅	4-ClC ₆ H ₄	6a (79)
2	H	CH(CH ₃) ₂	C ₆ H ₅	6b (82)
3	H	CH(CH ₃) ₂	4-ClC ₆ H ₄	6c (85)

^a All reactions were carried out with the ratio of LDA : imine : *N*-tosylimine = 1.6 : 1.5 : 0.75. ^b Isolated yields based on the *N*-tosylimine.

**Fig. 1** ORTEP drawing of **5g**.

with hydrogen at the α -position afforded the carbanion, however, no carbanion formed if sodium hydride was used as base. Butyllithium reacted with the C=N double bond.^{2c} With lithium bis(trimethylsilyl)amide as base, the product is complex.⁹

If NaBH₄-MeOH instead of oxalic acid was used, β -amino aldimines **4** can also be reduced smoothly to give 1,3-diamines in good yield (Scheme 2, Table 2). The NMR of **6** showed that

**Scheme 2**

there were two *NHT*s peaks with a ratio of around 1 : 1. Thus the epimerization should take place during the reduction reaction because both reactions have the same intermediate **4** and hydrolysis of **4** gave the products with *syn* : *anti* ratio of 40 : 60 to 8 : 92.

In conclusion, new and one-pot procedures have been developed to prepare β -aminoaldehydes and 1,3-diamines from α -carbanions of imines in a convenient way. The usefulness of α -carbanions of imines in organic synthesis is also demon-

strated. The investigation on the reactions of α -carbanions of imines with other kinds of nucleophiles and the asymmetric version of them are in progress.

Experimental

General

All the reactions were performed under a dry argon atmosphere. The commercially available reagents were used without further purification. THF was freshly distilled under nitrogen from a purple solution of sodium and benzophenone. ¹H-NMR spectra were recorded on a Bruker AMX-300 (300 MHz) spectrometer and the chemical shifts were referenced to CHCl₃ (δ 7.27) in CDCl₃, IR spectra were recorded in neat solutions, and measured in cm⁻¹, using a Shimadzu IR-440 infrared spectrophotometer. Mass spectra were taken using HP5989A. Elemental analyses were performed on a Foss-Heraeus Vario EL instrument.

General procedure for the preparation of β -aminoaldehydes **5**

To a solution of aldimine **1** (1.5 mmol) in THF (5 mL) was added LDA (1 mL, 1.6 M in THF) at 0 °C under argon, the resulting mixture was stirred for 2 h, then was cooled to -78 °C and *N*-tosylimine **2** (0.75 mmol) was added. The mixture was stirred at this temperature for 7 h. A solution of oxalic acid (144 mg, 1.6 mmol) in water (2 mL) was added dropwise and the mixture was stirred for 12 h. The mixture was extracted with Et₂O (10 mL \times 3). The organic layer was combined, washed with brine (10 mL \times 2) and dried over Na₂SO₄. The solvent was removed in vacuum and the residue was purified by flash column chromatography by using petroleum ether (60–90 °C) and ethyl acetate (3 : 1) as the eluent to obtain the corresponding β -aminoaldehydes **5** as a *syn-anti* mixture.

3-(*N*-Tosylamino)-3-(4'-chlorophenyl)-2-methylpropanal **5a**.

¹H-NMR: δ (CDCl₃-Me₄Si) *syn-5a*: 1.09 (d, J = 7.4 Hz, 3H), 2.37 (s, 3H), 2.76 (m, 1H), 4.69 (dd, J = 6.1 and 9.1 Hz, 1H), 6.05 (d, J = 9.4 Hz, 1H), 6.94 (m, 2H), 7.09 (m, 4H), 7.46 (m, 2H), 9.53 (d, J = 1.2 Hz, 1H); *anti-5a*: 0.96 (d, J = 7.2 Hz, 3H), 2.35 (s, 3H), 2.76 (m, 1H), 4.52 (dd, J = 8.4 and 9.0 Hz, 1H), 6.16 (d, J = 9.1 Hz, 1H), 6.94 (m, 2H), 7.09 (m, 4H), 7.46 (m, 2H), 9.62 (d, J = 2.7 Hz, 1H); MS: m/z (%) 351 (M⁺, 1.98), 260 (90), 91 (100); IR (cm⁻¹) 3273, 2977, 2925, 2720, 1721, 1598, 1495, 1455; Anal. Calcd C₁₇H₁₈ClNO₃S: C, 58.03; H, 5.16; N, 3.98. Found: C, 57.76; H, 5.43; N, 3.96%.

3-(*N*-Tosylamino)-3-(4'-chlorophenyl)-2,2-dimethylpropanal **5b**.

¹H-NMR: δ (CDCl₃-Me₄Si) 1.01 (s, 3H), 1.05 (s, 3H), 2.33 (s, 3H), 4.46 (d, J = 9.9 Hz, 1H), 6.21 (d, J = 9.9 Hz, 1H), 6.85 (d, J = 8.3 Hz, 2H), 7.02 (d, J = 8.6 Hz, 4H), 7.41 (d, J = 8.3 Hz, 2H), 9.49 (s, 1H); MS: m/z (%) 365 (M⁺, 0.25), 293 (100); IR (cm⁻¹) 3263, 2928, 2843, 2710, 1735, 1598, 1461, 1437; Anal. Calcd. C₁₈H₂₀ClNO₃S: C, 59.09; H, 5.51; N, 3.83. Found: C, 59.04; H, 5.78; N, 3.57%.

3-(*N*-Tosylamino)-3-phenyl-2-ethylpropanal **5c**.

¹H-NMR: δ (CDCl₃-Me₄Si) *syn-5c*: 0.92 (t, J = 7.3 Hz, 3H), 1.37–1.60 (m, 2H), 2.32 (s, 3H), 2.58 (m, 1H), 4.59 (dd, J = 8.0 and 8.9 Hz, 1H), 5.55 (d, J = 8.9 Hz, 1H), 6.95–7.11 (m, 7H), 7.45 (m, 2H), 9.46 (d, J = 2.3 Hz, 1H); *anti-5c*: 0.89 (t, J = 7.8 Hz, 3H), 1.37–1.60 (m, 2H), 2.30 (s, 3H), 2.58 (m, 1H), 4.59 (dd, J = 8.0 and 8.9 Hz, 1H), 5.72 (d, J = 9.5 Hz, 1H), 6.95–7.11 (m, 7H), 7.45 (m, 2H), 9.56 (d, J = 3.2 Hz, 1H); MS: m/z (%) 314 (M⁺ - 17, 6.23), 260 (100); IR (cm⁻¹) 3240, 3028, 2968, 2697, 1726, 1598, 1494, 1455, 1385, 1324; Anal. Calcd. C₁₈H₂₁NO₃S: C, 65.23; H, 6.38; N, 4.22. Found: C, 65.03; H, 6.29; N, 4.35%.

3-(*N*-Tosylamino)-3-(4'-methoxyphenyl)-2-ethylpropanal **5d**.

¹H-NMR: δ (CDCl₃-Me₄Si) *syn-5d*: 0.90 (t, J = 7.4 Hz, 3H),

1.33 (m, 1H), 1.51 (m, 1H), 2.33 (s, 3H), 2.54 (m, 1H), 3.71 (s, 3H), 4.54 (dd, $J = 9.1$ and 8.8 Hz, 1H), 5.77 (d, $J = 9.0$ Hz, 1H), 6.63 (m, 2H), 6.88 (m, 2H), 7.06 (m, 2H), 7.46 (m, 2H), 9.44 (d, $J = 2.3$ Hz, 1H); *anti*-**5d**: 0.84 (t, $J = 7.8$ Hz, 3H), 1.33 (m, 1H), 1.51 (m, 1H), 2.31 (s, 3H), 2.54 (m, 1H), 3.72 (s, 3H), 4.54 (dd, $J = 9.1$ and 8.8 Hz, 1H), 5.90 (d, $J = 8.9$ Hz, 1H), 6.63 (m, 2H), 6.88 (m, 2H), 7.06 (m, 2H), 7.46 (m, 2H), 9.58 (d, $J = 3.9$ Hz, 1H); MS: m/z (%) 290 ($M^+ - 71$, 100); IR (cm^{-1}) 3261, 2960, 2923, 2710, 1726, 1613, 1515, 1461; Anal. Calcd. $\text{C}_{19}\text{H}_{23}\text{NO}_4\text{S}$: C, 63.14; H, 6.41; N, 3.88. Found: C, 62.81; H, 6.50; N, 3.57%.

3-(*N*-Tosylamino)-3-(4'-chlorophenyl)-2-ethylpropanal 5e. $^1\text{H-NMR}$: δ ($\text{CDCl}_3\text{-Me}_4\text{Si}$) *syn*-**5e**: 0.89 (m, 3H), 1.36–1.67 (m, 2H), 2.34 (s, 3H), 2.55 (m, 1H), 4.57 (m, 1H), 5.85–6.06 (m, 1H), 6.91–7.06 (m, 6H), 7.43 (m, 2H), 9.44 (d, $J = 2.1$ Hz, 1H); *anti*-**5e**: δ 0.89 (m, 3H), 1.36–1.67 (m, 2H), 2.33 (s, 3H), 2.55 (m, 1H), 4.57 (m, 1H), 5.85–6.06 (m, 1H), 6.91–7.06 (m, 6H), 7.43 (m, 2H), 9.56 (d, $J = 3.5$ Hz, 1H); MS: m/z (%) 294 ($M^+ - 71$, 43), 155 (30), 91 (100); IR (cm^{-1}) 3261, 2960, 2923, 2710, 1726, 1613, 1515, 1461, 1323; Anal. Calcd. $\text{C}_{18}\text{H}_{20}\text{ClNO}_3\text{S}$: C, 59.08; H, 5.50; N, 3.82. Found: C, 59.12; H, 5.43; N, 3.67%.

3-(*N*-Tosylamino)-3-(4'-methoxyphenyl)-2-propylpropanal 5f. $^1\text{H-NMR}$: δ ($\text{CDCl}_3\text{-Me}_4\text{Si}$) *syn*-**5f**: 0.85 (m, 3H), 1.31–1.53 (m, 4H), 2.33 (s, 3H), 2.69 (m, 1H), 4.58 (m, 1H), 5.49 (d, $J = 8.5$ Hz, 1H), 6.94–7.14 (m, 7H), 7.47 (m, 2H), 9.47 (d, $J = 2.3$ Hz, 1H); *anti*-**5f**: 0.85 (m, 3H), 1.31–1.53 (m, 4H), 2.32 (s, 3H), 2.69 (m, 1H), 4.58 (m, 1H), 5.67 (d, $J = 9.4$ Hz, 1H), 6.94–7.14 (m, 7H), 7.47 (m, 2H), 9.54 (d, $J = 3.0$ Hz, 1H); MS: m/z (%) 328 ($M^+ - 17$, 1.48), 260 (32), 155 (34), 91 (100), 77 (10); IR (cm^{-1}): 3269, 2959, 2870, 2710, 1729, 1457, 1381, 1322, 1289; Anal. Calcd. $\text{C}_{19}\text{H}_{23}\text{NO}_3\text{S}$: C, 66.06; H, 6.71; N, 4.05. Found: C, 65.86; H, 6.80; N, 3.94%.

3-(*N*-Tosylamino)-3-phenyl-2-isopropylpropanal 5g. $^1\text{H-NMR}$: δ ($\text{CDCl}_3\text{-Me}_4\text{Si}$) *syn*-**5g**: 1.01 (m, 6H), 1.80 (m, 1H), 2.32 (s, 3H), 2.54 (m, 1H), 4.71 (dd, $J = 9.1$ and 8.8 Hz, 1H), 5.58 (d, $J = 8.8$ Hz, 1H), 7.01 (m, 7H), 7.42 (m, 2H), 9.45 (d, $J = 3.4$ Hz, 1H); *anti*-**5g**: 1.01 (m, 6H), 1.80 (m, 1H), 2.29 (s, 3H), 2.54 (m, 1H), 4.82 (dd, $J = 8.9$ and 8.0 Hz, 1H), 5.96 (d, $J = 9.4$ Hz, 1H), 7.01 (m, 7H), 7.42 (m, 2H), 9.74 (d, $J = 3.4$ Hz, 1H); MS: m/z (%) 314 ($M^+ - 31$, 2.23), 260 (70), 91 (100); IR (cm^{-1}) 3247, 2966, 2770, 1708, 1600, 1490, 1456, 1329; Anal. Calcd. $\text{C}_{19}\text{H}_{23}\text{NO}_3\text{S}$: C, 66.06; H, 6.71; N, 4.05. Found: C, 65.80; H, 6.66; N, 3.76%.

3-(*N*-Tosylamino)-3-(4'-methoxyphenyl)-2-isopropylpropanal 5h. $^1\text{H-NMR}$: δ ($\text{CDCl}_3\text{-Me}_4\text{Si}$) *syn*-**5h**: 1.00 (m, 6H), 1.75 (m, 1H), 2.33 (s, 3H), 2.53 (m, 1H), 3.70 (s, 3H), 4.69 (dd, $J = 8.7$ and 9.0 Hz, 1H), 6.00 (d, $J = 9.1$ Hz, 1H), 6.60 (m, 2H), 6.89 (m, 2H), 7.03 (m, 2H), 7.41 (m, 2H), 9.44 (d, $J = 3.4$ Hz, 1H); *anti*-**5h**: 0.97 (d, $J = 7.7$ Hz, 3H), 0.99 (d, $J = 7.7$ Hz, 3H), 1.75 (m, 1H), 2.31 (s, 3H), 2.53 (m, 1H), 3.72 (s, 3H), 4.78 (dd, $J = 8.9$ and 9.1 Hz, 1H), 6.00 (d, $J = 9.1$ Hz, 1H), 6.60 (m, 2H), 6.89 (m, 2H), 7.03 (m, 2H), 7.41 (m, 2H), 9.76 (d, $J = 4.0$ Hz, 1H); MS: m/z (%) 314 ($M^+ - 31$, 2.23), 260 (70), 155 (43), 91 (100); IR (cm^{-1}) 3247, 2966, 2770, 1708, 1600, 1490, 1456, 1329; Anal. Calcd. $\text{C}_{20}\text{H}_{25}\text{NO}_4\text{S}$: C, 63.97; H, 6.71; N, 3.73. Found: C, 64.14; H, 7.03; N, 3.50%.

3-(*N*-Tosylamino)-3-(4'-chlorophenyl)-2-isopropylpropanal 5i. $^1\text{H-NMR}$: δ ($\text{CDCl}_3\text{-Me}_4\text{Si}$) *syn*-**5i**: 1.00 (m, 6H), 1.76 (m, 1H), 2.35 (s, 3H), 2.51 (m, 1H), 4.67 (dd, $J = 9.5$ and 8.2 Hz, 1H), 5.90 (d, $J = 8.7$ Hz, 1H), 6.91 (m, 2H), 7.03 (m, 4H), 7.39 (m, 2H), 9.45 (d, $J = 3.2$ Hz, 1H); *anti*-**5i**: 1.00 (m, 6H), 1.76 (m, 1H), 2.34 (s, 3H), 2.51 (m, 1H), 4.80 (dd, $J = 9.2$ and 8.4 Hz, 1H), 6.21 (d, $J = 9.6$ Hz, 1H), 6.91 (m, 2H), 7.03 (m, 4H), 7.39 (m, 2H), 9.74 (d, $J = 3.7$ Hz, 1H); MS: m/z (%) 290 ($M^+ - 85$, 2.23), 155 (22), 92 (22), 91 (100); IR (cm^{-1}) 3249, 2961, 2710, 1720, 1612, 1597, 1561; Anal. Calcd. $\text{C}_{19}\text{H}_{22}\text{ClNO}_3\text{S}$: C, 60.07; H, 5.83; N, 3.69. Found: C, 59.82; H, 5.82; N, 3.59%.

General procedure for the preparation of 1,3-diamines 6

To a solution of aldimine **1** (1.5 mmol) in THF (5 mL) was added LDA (1 mL, 1.6 M in THF) at 0°C under argon, the resulting mixture was stirred for 2 h, then was cooled to -78°C , and *N*-tosylimine (**2**, 0.75 mmol) was added, the mixture was stirred at this temperature for 7 h. Then NaBH_4 (10 mmol) in MeOH (4 mL) was added and the temperature was allowed to rise to room temperature. After stirring for 3 h at room temperature, HCl (15 mL, 2 M) was added and the mixture was extracted with Et_2O (3×20 mL). Subsequently, the aqueous phase was basified by the addition of NH_3 , and the 1,3-diamines **6** were extracted with CH_2Cl_2 (3×20 mL). The combined organic phase was dried over Na_2SO_4 . The solvent was removed in vacuum and the residue was purified by preparative TLC with a mixture of light petroleum ($60\text{--}90^\circ\text{C}$) and ethyl acetate (3 : 1) as the eluent to give pure product **6** as a *syn-anti* mixture.

3-(*N*-Tosylamino)-3-(4'-chlorophenyl)-2-ethyl-*N*-(*tert*-butyl)propylamine 6a. $^1\text{H-NMR}$: δ ($\text{CDCl}_3\text{-Me}_4\text{Si}$) *syn*-**6a**: 0.89 (t, $J = 7.6$ Hz, 3H), 1.13 (s, 9H), 0.90–1.68 (m, 4H), 2.36 (s, 3H), 2.42 (d, $J = 10.3$ Hz, 1H), 2.56 (m, 1H), 4.61 (d, $J = 4.1$ Hz, 1H), 7.05 (m, 6H), 7.45 (m, 2H); *anti*-**6a**: 0.79 (t, $J = 7.3$ Hz, 3H), 1.15 (s, 9H), 0.90–1.68 (m, 4H), 2.35 (s, 3H), 2.56 (m, 1H), 2.74 (dd, $J = 1.8$ and 12.0 Hz, 1H), 4.36 (d, $J = 6.8$ Hz, 1H), 7.05 (6H, m), 7.45 (2H, m); MS: m/z (%) 423 ($M^+ + 1$, 11.85), 407 (32.24), 91 (27.28), 86 (100); IR (cm^{-1}) 3302, 2967, 2876, 1599, 1492, 1323, 1290, 1215; Anal. Calcd. $\text{C}_{22}\text{H}_{31}\text{ClN}_2\text{O}_2\text{S}$: C, 62.46; H, 7.39; N, 6.62. Found: C, 62.43; H, 7.52; N, 6.43%.

3-(*N*-Tosylamino)-3-phenyl-2-isopropyl-*N*-(*tert*-butyl)propylamine 6b. $^1\text{H-NMR}$: δ ($\text{CDCl}_3\text{-Me}_4\text{Si}$) *syn*-**6b**: 0.87 (d, $J = 6.4$ Hz, 3H), 0.95 (d, $J = 6.3$ Hz, 3H), 1.14 (s, 9H), 1.27–1.53 (m, 3H), 2.33 (s, 3H), 4.70 (d, $J = 3.9$ Hz, 1H), 7.03–7.11 (m, 7H), 7.43 (m, 2H); *anti*-**6b**: 0.51 (d, $J = 6.6$ Hz, 3H), 0.83 (d, $J = 7.0$ Hz, 3H), 1.15 (s, 9H), 1.27–1.53 (m, 3H), 2.32 (s, 3H), 4.59 (d, $J = 7.0$ Hz, 1H), 7.03–7.11 (m, 7H), 7.43 (m, 2H); MS: m/z (%) 387 ($M^+ - 15$, 20.82), 260 (11.09), 91 (100); IR (cm^{-1}) 3298, 2958, 2872, 1601, 1493, 1454, 1346; Anal. Calcd. $\text{C}_{23}\text{H}_{34}\text{N}_2\text{O}_2\text{S}$: C, 68.61; H, 8.51; N, 6.96. Found: C, 68.33; H, 8.36; N, 6.56%.

3-(*N*-Tosylamino)-3-(4'-chlorophenyl)-2-isopropyl-*N*-(*tert*-butyl)propylamine 6c. $^1\text{H-NMR}$: δ ($\text{CDCl}_3\text{-Me}_4\text{Si}$) *syn*-**6c**: 0.85 (d, $J = 6.8$ Hz, 3H), 0.94 (d, $J = 6.8$ Hz, 3H), 1.17 (s, 9H), 1.14–1.58 (m, 4H), 2.53 (s, 3H), 2.39–2.55 (m, 1H), 2.71 (t, $J = 10.3$ Hz, 1H), 4.69 (d, $J = 3.8$ Hz, 1H), 7.03 (m, 6H), 7.41 (m, 2H); *anti*-**6c**: 0.52 (d, $J = 6.7$ Hz, 3H), 0.81 (d, $J = 6.7$ Hz, 3H), 1.13 (s, 9H), 1.14–1.58 (m, 4H), 2.34 (s, 3H), 2.39–2.55 (m, 1H), 2.71 (t, $J = 10.3$ Hz, 1H), 4.50 (d, $J = 7.6$ Hz, 1H), 7.03 (m, 6H), 7.41 (m, 2H); MS: m/z (%) 437 ($M^+ + 1$, 17.02), 421 (23.03), 86 (100); IR (cm^{-1}) 3303, 2957, 2872, 1598, 1491, 1412, 1312, 1248, 1214. Anal. Calcd. $\text{C}_{23}\text{H}_{33}\text{ClN}_2\text{O}_2\text{S}$: C, 63.21; H, 7.61; N, 6.40. Found: C, 63.08; H, 7.67; N, 6.16%.

X-Ray diffraction analysis data of 3-*N*-tosylamino-3-phenyl-2-isopropylpropanal 5g †

Chemical formula: $\text{C}_{19}\text{H}_{23}\text{NO}_3\text{S}$; formula weight: 345.44; crystal system: triclinic; unit-cell dimensions and volume with estimated standard deviations: cell length a : 9.4619(11); cell length b : 9.8687(12); cell length c : 10.1330(12); cell angle alpha: 100.097(2); cell angle beta: 99.529(2); cell angle gamma: 91.423(2); cell volume: 917.26(19); data collection temperature: 293 K; No. of formula units in unit cell (Z): 2; linear absorption coefficient (μ): 0.192 mm^{-1} ; Number of reflections measured and/or number of independent reflections, 3371 (R_{int}) = 0.0401; Final R indices [$I > 2\sigma(I)$]: $R_1 = 0.0406$, $wR_2 = 0.0507$.

† CCDC reference number 177332. See <http://www.rsc.org/suppdata/p1/b2/b200198e/> for crystallographic files in .cif or other electronic format.

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References

- 1 For some reviews see: (a) J. P. Adams and G. Robertson, *Contemp. Org. Synth.*, 1997, **4**, 183; (b) R. Bloch, *Chem. Rev.*, 1998, **98**, 1407; (c) S. Kobayashi and H. Ishitani, *Chem. Rev.*, 1999, **99**, 1069; (d) J. P. Adams, *J. Chem. Soc., Perkin Trans. 1*, 2000, 125.
- 2 (a) A. H. Li, Y. G. Zhou, L. X. Dai, X. L. Hou, L. J. Xia and L. Lin, *J. Org. Chem.*, 1998, **63**, 4338; (b) V. K. Aggarwal, M. Ferrara, C. J. O'Brien, A. Thoompson, R. V. H. Jones and R. Fieldhouse, *J. Chem. Soc., Perkin Trans. 1*, 2001, 1635; (c) V. K. Aggarwal, *Synlett*, 1998, 329; (d) F. A. Davis, H. L. P. Zhou, T. Fang, G. V. Reddy and Y. Zhang, *J. Org. Chem.*, 1999, **64**, 7559; (e) S. E. Denmark, *Chem. Commun.*, 1996, 999; (f) T. Takahashi, Y. Liu, C. Xi and S. Huo, *Chem. Commun.*, 2001, 31; (g) X. L. Hou, X.-L. Zheng and L.-X. Dai, *Tetrahedron Lett.*, 1998, **39**, 6949; (h) S. Kobayashi, H. Ishitani, Y. Yamashita, M. Ueno and H. Shimizu, *Tetrahedron*, 2001, **57**, 861; (i) K. A. Jørgensen, *Angew. Chem., Int. Ed.*, 2000, **39**, 3558; (j) C. T. Qian, L. C. Wang and R. F. Chen, *Chin. J. Chem.*, 2001, **19**, 419.
- 3 (a) C. V. Stevens and N. G. kimpe, *Tetrahedron Lett.*, 1994, **35**, 3763; (b) S. Fustero, M. G. de la Torre, V. Jofré, R. P. Carlón, A. Navarro and A. S. Fuentes, *J. Org. Chem.*, 1998, **63**, 8825; (c) C. Gimarelli and G. Palmier, *Tetrahedron*, 1998, **54**, 15711; (d) J. L. G. Ruano, A. Lorente and J. H. R. Ramos, *Tetrahedron: Asymmetry*, 1998, **9**, 2437; (e) S. J. Veenstra and S. S. Kinderman, *Synlett*, 2001, 1109.
- 4 M. B. Smith and J. March, *March's Advanced Organic Chemistry: Reactions, Mechanisms, and Structure*, 5th edn., Wiley, New York, 2001.
- 5 (a) D. K. Wang, L. X. Dai and X. L. Hou, *Tetrahedron Lett.*, 1995, **36**, 8649; (b) A. H. Li, L. X. Dai and X. L. Hou, *J. Chem. Soc., Perkin Trans. 1*, 1996, 867; (c) A. H. Li, L. X. Dai and X. L. Hou, *J. Chem. Soc., Perkin Trans. 1*, 1996, 2725; (d) A. H. Li, L. X. Dai, X. L. Hou and M. B. Chen, *J. Org. Chem.*, 1996, **61**, 4641; (e) A. H. Li, Y. G. Zhou, L. X. Dai, X. L. Hou, L. J. Xia and L. Lin, *Angew. Chem., Int. Ed. Engl.*, 1997, **36**, 1317; (f) X. L. Hou, X. F. Yang, L. X. Dai and X. F. Chen, *Chem. Commun.*, 1998, 747; (g) D. K. Wang, Y. G. Zhou, Y. Tang, X. L. Hou and L. X. Dai, *J. Org. Chem.*, 1999, **64**, 4233; (h) X. F. Yang, X. L. Hou and L. X. Dai, *Tetrahedron Lett.*, 2000, **41**, 4431.
- 6 (a) S. B. Davies and M. A. McKervey, *Tetrahedron Lett.*, 1999, **40**, 1229; (b) B. Merla, M. Arend and N. Risch, *Synlett*, 1997, 177.
- 7 R. W. M. Aben, R. Smit and J. W. Scheeren, *J. Org. Chem.*, 1987, **52**, 365.
- 8 *Enamines: Synthesis, structure and reactions*, ed. A. G. Cook, 2nd edn., Dekker: New York, 1988.
- 9 D. J. Berrisford, *Angew. Chem., Int. Ed. Engl.*, 1995, **34**, 178.